Error Analysis

"To err is human; to describe the error properly is sublime." - Cliff Swartz (1999)

Today

- A discussion about a laboratory topic: error analysis
- Your learning of this:
- 1. The assignment
- 2. Using error analysis in an experiment
- 3. This talk
- 4. A test (administered via computer)



Coming Next Week...



- We will begin the *Waves Quarter* on Oscillations, Sound and Light.
- For Monday, please read Sections 14.1 through 14.3 of Knight.
- There is a Pre-Class Quiz (Waves #1) on Chapter 14 due Monday morning on www.masteringphysics.com.

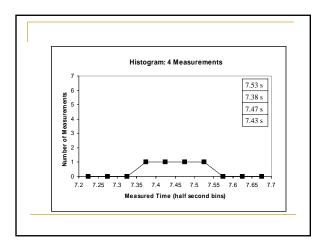
Two Kinds of Statements

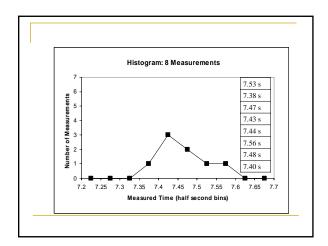
- 1. Exact
 - 2 + 3 = 5 (math)
 - $K = \frac{1}{2} mv^2$ (definition)
- 2. Approximate
 - F_{spring} = -kx (*any* physical law)
 - g = 9.80 m/s² (*all* numerical measures of the universe)

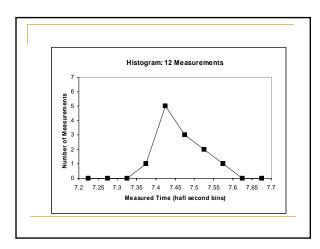
Today: approximate statements

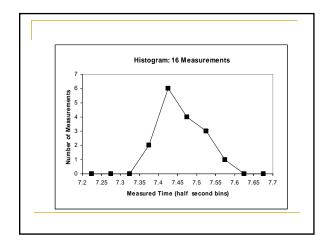
The t_5 data

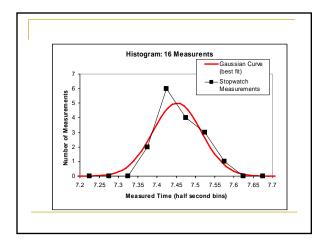
7.53 s 7.38 s 7.47 s 7.43 s

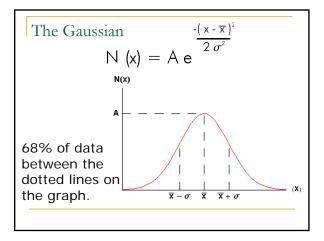


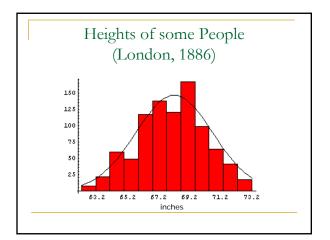


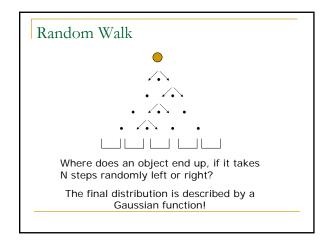


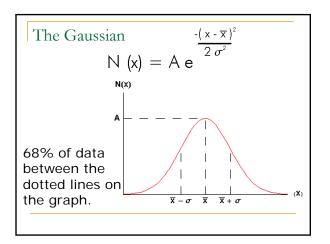






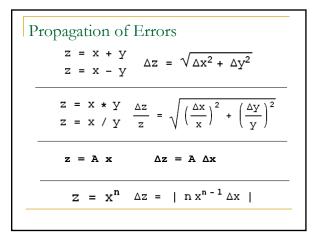






	The t_5 data
7.53 s ± 0.06 s	Numerically:
7.38 s ± 0.06 s	$\bar{t}_{5,est} = 7.45250 s$
7.47 s ± 0.06 s	$\sigma_{est} = 0.0634429 s$
7.43 s ± 0.06 s	$\sigma_{est} = 0.06 s$





- Repeated **n** times
- Each individual measurement has an error of precision Δx

$$\Delta \bar{\mathbf{x}}_{est} = \frac{\Delta \mathbf{x}}{\sqrt{n}}$$

Significant Figures

Discussed in Section 1.9 of Knight Ch.1

- Rules for significant figures follow from error
 - propagation
 Assume error in a quoted value is half the value of the last digit.
 - Errors should be quoted to 1 or 2 significant figures
 - Error should be in final displayed digit in number.
- Example: If a calculated result is: (7.056 ± 0.705) m, it is better to report:

(7.1 ± 0.7) m.